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Kinetic Resolution of Secondary Alcohols Mediated by Rabbit Gastric Lipase

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Abstract: Secondary benzylic alcohols were kinetically resolved by rabbit gastric lipase-mediated acylation with isopropenyl acetate. Among all the substrates tested, a high enantioselectivity was observed only for $1-\{2-(6-methox)naphthyl]$ ethanol **3a** (E>500). Allylic alcohols **14a** and **15a** were also efficiently resolved (E = 26 and 51 respectively). © 1997 Elsevier Science Ltd. All rights reserved.

We recently described the palladium-catalyzed nucleophilic substitution of enantiomerically pure esters of alcohols **1a-3a** by sodium dimethyl malonate.¹ Alcohols **1a** and **2a** are commercially available, in racemic and both enantiomeric forms. Racemic **3a** is easily prepared by hydride reduction of the corresponding ketone.² We wish to report our results concerning the preparation of enantiomerically pure **3a** and **3b**,³ and various others optically active 1-arylethanols, by kinetic resolution mediated by rabbit gastric lipase (RGL).⁴ We already used RGL in kinetic resolution of 2-hydroxyalkyldiphenylphosphines⁵ and axially chiral cycloalkylidenethanols.⁶

$\begin{array}{ccc} 3 & Ar = 2-(6-methoxy)naphthyl & 9\\ 4 & Ar = phenyl & 10\\ 5 & Ar = 2-methoxyphenyl & 11 \end{array}$	Ar = 4-bromophenyl Ar = ferrocenyl $Ar = 3$ -quinolyl Ar = 4-quinolyl Ar = 4-isoquinolyl
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Racemic 1-[2-(6-methoxynaphthyl)]ethanol 3a (1g, 5 mmol) and isopropenyl acetate (1.5 g, 15 mmol) were dissolved in 30 mL ether and 200 mg of RGL were added. The resulting suspension was stirred at room temperature for 21.5 h and then filtered through Celite. After evaporation of the solvent, the residue was subjected to chromatography (silica gel, eluent = hexane / ethyl acetate 80:20) to give (*R*)-3b (0.51 g, 2.1 mmol, 42%), ee > 99.5% and (*S*)-3a (0.41 g, 2.0 mmol, 41%), ee = 78.5%. Enantiomeric excesses were determined by HPLC on a CHIRACEL OD-H column. The substrate conversion and the enantioselectivity factor E were calculated⁷ to be 44% and > 500 respectively. This high value of E is superior to the reported value in the literature (E = 78),^{3b} and to the results we obtained with some commercially available lipases. Saponification (KOH, MeOH) of (*R*)-3b gave enantiomerically pure (*R*)-3a.

Encouraged by this satisfactory result, we attempted the acylation of 3a by succinic anhydride,⁹ in order to avoid the separation of 3a and 3b by column chromatography. However, although the enzymatic activity was retained, the enantioselectivity of this resolution was very low (E = 2).

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Some other 1-arylethanols were tested in the acetylation by isopropenyl acetate: experimental conditions were the same as described above for 3a. Results are collected in the Table.

These kinetic resolutions were performed in order to determine the factor E. Enantiomeric excesses of remaining substrates were not optimized. The present study demonstrate that RGL shown a high specificity for alcohol 3a. For all other substrates, the E values were in the range 1.1 to 23. Utilization of RGL in the case of 1a, 2a, 4a, 9a and 11a does not improve the results described in the literature.¹⁰

1a and **2a** showed moderate enantioselectivity with a very low enzymatic activity for the former (entries 1 and 2). The large difference on substrates **2a** and **3a** should be noted: the introduction of a methoxy substituent in the 6-position of **2a** enhances the enantioselectivity by a factor 20 at least. The enantiomers of 1-acenaphthenol **13a** were both rapidly esterified in the same conditions (entry 4).

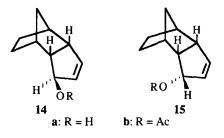
1-Phenylethanol 4a was poorly resolved by RGL (entry 5) and introduction of a methoxy substituent in meta (6a) or para (7a) position resulted in a moderate enhancement of the enantioselectivity (entries 6 and 7). The results of entries 5 to 7 showed that the large difference observed between the enantioselectivities on 2a and 3a is not only due to an electronic effect. 1-(2-methoxyphenyl)ethanol 5a was recovered unchanged after 5 days of reaction, maybe because of a strong hydrogen bond stabilization of the substrate.

The introduction of a para-bromo substituent is beneficial (entry 8), and the recovered 1-(4-bromo)phenylethanol 8a was enantiomerically pure at 60% conversion.

Both enantiomers of 1-ferrocenylethanol 9a were acetylated with very similar rates and both the alcohol 9a and the acetate 9b were obtained with very low ee values (entry 9).

We investigated three heteroaromatic substrates **10a-12a**: (+)-1-(3-quinolyl)ethanol **10a** was previously obtained by asymmetric hydride reduction of 3-acetylquinoline,¹¹ but the enantioselectivity of the reaction was unknown. 1-(4-Quinolyl)ethanol **11a** was recently resolved via an immobilized lipase-catalyzed acetylation. ^{10d} To the best of our knowlegde, 1-(4-isoquinolyl)ethanol **12a** was described only in racemic form. Substrate **10a** gave a comparable result to the similarly substituted naphthalene derivative **2a** (entry 10), whereas **11a** and **12a** (entries 11 and 12) were slowly acetylated, with a lower enantioselectivity than **1a**. Introduction of a nitrogen atom on the aromatic moiety has little influence on this resolution.

Finally, we examined the kinetic resolution of allylic alcohols **14a** and **15a**.¹² Commercially available PPL and CCL lipases were not efficient in the acetylation of (\pm) -**14a**: the enzymatic activities were low (conversion of 17% and 33% respectively in 3 days) as well as the enantioselectivities (E = 5 and 3). RGL gave the best result for this substrate: (+)-**14a** (ee = 70%) and (-)-**14b** (ee = 85%) were obtained after 44h at 20°C in cyclohexane (c = 0.45, E = 26). The acetylation of the endo isomer (\pm)-**15a** was slower since after 5 days at 30°C (no acetate was detected after two days of reaction at 20°C), the conversion reached about the same value (c = 0.44). However, the enantioselectivity was enhanced: alcohol (+)-**15a** and acetate (-)-**15b** were isolated with 71% and 92% ee respectively (E = 51).¹³



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Entry		Substrate	time (h)	ee alcohol (%) a	ee acetate (%) a	c (%) ^b	Еb		
1	la	C	96	13	85	13.5	14		
2	2a	ОН	19	59 c	85.5	41	23		
3	3a	мео	21.5	78.5	>99.5	44	>500		
4	13a		24	80.5	2.5 d	97	2		
5	4 a	MeQ.	18	54	43.5 d	55.5	4		
6	ба	С	18	92	63 d	59.5	14		
7	7a	MeO-	18	95 c	36	72.5	7		
8	8a		18	>99.5	63 d	61	20		
9	9a	OH COH	18	6 c	4	60	1.1		
10	10a	ОН	15	82.5	74.5 d	52.5	17		
11	11a		17	7	68.5	9	6		
12	12a	OH N	96	17 ¢	59 e	22	5		

Table

^a Determined by HPLC unless otherwise noted. ^b The substrate conversion c and the enantioselectivity factor E were calculated⁷ from standard equations. ^c Determined by HPLC after acetylation (Ac₂O, catalytic DMAP, Et₃N). ^d Determined by HPLC after saponification (KOH, MeOH). ^c Determined by ¹H NMR in the presence of Eu(hfc)₃.

In conclusion, enantioselectivities for kinetic resolution of a number of secondary alcohols have been determined in lipase-mediated acetylation. Rabbit gastric lipase was found to be highly superior in terms of enantioselection to common commercially available lipases in two cases (3a and 14a). In the case of bicyclic aromatic substrates, influence of steric factors in enantioselection predominates since β -substituted aryl methyl carbinols showed higher selectivity than α -isomers.

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